# Studies on annulated 1,4-benzothiazines and 1,5-benzothiazepines. IX. Imidazo[2,1-d][1,5]benzothiazepines: synthesis and in vitro benzodiazepine receptor affinity

V Ambrogi<sup>1</sup>, G Grandolini<sup>1\*</sup>, L Perioli<sup>1</sup>, L Giusti<sup>2</sup>, A Lucacchini<sup>2</sup>, C Martini<sup>2</sup>

<sup>1</sup>Istituto di Chimica Farmaceutica e Tecnica Farmaceutica, Università di Perugia, Via del Liceo, 1, 06123 Perugia; 
<sup>2</sup>Istituto Policattedra di Discipline Biologiche, Università di Pisa, Via Bonanno, 6, 56100 Pisa, Italy

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Summary — The synthesis of three series of 1- and 2-substituted imidazo[2,1-d][1,5]benzothiazepines is accomplished starting from 1,5-benzothiazepin-4-ones. All the synthesized compounds were evaluated for their affinity for the benzodiazepine receptor, testing their ability to displace [3H]Flunitrazepam from bovine brain membrane protein. A few of the tested compounds showed good affinity, in particular compound 9a ( $K_i = 43.00 \text{ nM}$ ). The GABA-ratio of the active compounds suggests an antagonist or partial agonist activity. The data obtained allow us to draw some comments on the structure—activity relationships.

imidazo[2,1-d][1,5]benzothiazepine / benzodiazepine receptor affinity / structure-activity relationship

#### Introduction

Since the discovery of benzodiazepine receptor a number of ligands with different structures have been found. Recently, we have synthesized some triazolo[3,4-d][1,5]benzothiazepines, which showed moderate affinity for the benzodiazepine receptor [1, 2]. With the aim of expanding available structure—activity relationship (SAR) correlations, we describe here the synthesis and the biological evaluation of some 1- or 2-substituted derivatives of imidazo[2,1-d]-[1,5]benzothiazepine, a heterocyclic ring system of which only a few derivatives are already known [3, 4].

Considering the biological importance of various imidazo derivatives [5–11] all the compounds described here were also screened as antiinflammatory and CNS agents. These results will be discussed in a forthcoming paper.

## Chemistry

The synthesis of 1-methyl derivatives 4a-g was achieved starting from the known 1,5-benzothiazepinthiones 2 [12, 13]. Compounds 4a-d were directly obtained by refluxing the corresponding thiones 2 with propargy-lamine hydrochloride, whereas the 5-aryl-substituted derivatives 4e-g and 12 were obtained through intermediates 3 [13] and 11 [1]. The direct reaction between 2-phenyl-2,3-dihydro-1,5-benzothiazepin-4-

(5H)-thione and propargylamine hydrochloride gave rise to 2-styrylbenzothiazole in agreement with our previous observations [1] (schemes 1 and 2).

2-Methyl or 2-phenyl derivatives 8a-g and 9a-g could be synthesized by two methods (scheme 1). According to the known method [3] (Method A), 4-amino-1,5-benzothiazepines 5 were used as intermediates for the synthesis of 8 and 9. Compounds 5 were prepared in a different manner to that indicated in the literature [3], starting from benzothiazepinones 1, which, after conversion into thiones 2, were reacted with ammonia in methanol. The final compounds were obtained by refluxing 4-aminobenzothiazepines 5 with either chloroacetone or phenacylbromide.

Alternatively (Method B), benzothiazepinones 1 were reacted with chloroacetone or phenacylbromide using the phase-transfer catalysis conditions (PTC). The N-substituted derivatives 6 and 7, obtained in good yields, were then refluxed with ammonium acetate in glacial acetic acid to give the desired compounds 8 and 9.

Considering the low yields obtained by *Method A*, the instability of compounds **5a-g** and the procedure length, we preferred to use *Method B* to obtain the compounds **8a-g** and **9a-g** and prepare the dehydro derivatives **15** and **16** (scheme 2).

The physical, spectral and analytical data for the synthesized compounds are given in the *Experimental* protocols and in tables I–V.

#### Scheme 1.

# Pharmacology

In two previous reports [1, 2] we reported the synthesis and the benzodiazepine receptor affinity of 1,5-benzothiazepines annulated with either a triazole or tetrazole ring, and 1,4-benzothiazines or 1,5-benzothiaze-

Scheme 2.

pines annulated with the triazine ring. Continuing our interest in this field we have now studied some new derivatives of imidazo [2,1-d][1,5] benzothiazepines.

The benzodiazepine receptor affinity of these compounds was measured by their ability to displace [ ${}^{3}H$ ]Flunitrazepam (at 0.2 nM,  $K_{d}$  = 1.8 nM) from its specific binding in bovine brain membranes. First a single concentration (10  $\mu$ M) of the tested compounds was examined and then, for the most active ones, the IC<sub>50</sub> values were calculated by log-probit plots.

From the  $IC_{50}$  values the  $K_i$  were calculated (table VI). These were used to define the benzodiazepine receptor affinity, and the GABA-ratio, which according to various authors [14–18] generally predicts the expected behaviour properties of a benzodiazepine receptor ligand.

# Results and discussion

The binding results showed the importance of the presence of the lipophilic phenyl substituent at the 2-position on the imidazole nucleus (compound 9a,  $K_i = 43$  nM). The replacement of the phenyl ring with a methyl group actually caused a decrease of two

Table I. Physical and chemical data of compounds 4a-g and 12.

$$X \longrightarrow \begin{pmatrix} H_3C & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\$$

Compound	X	R	Yield (%)	Mp (°C)	Crystallization solvent	Colour, crystal form	Formula	¹H-NMR, δ⁴
4a	Н	Н	33	114–115	b	_	$C_{12}H_{12}N_2S$	2.21 (s, 3H, CH <sub>3</sub> ), 2.33–3.70 (m, 4H, CH <sub>2</sub> CH <sub>2</sub> ), 6.82 (s, 1H, H-2), 7.16–7.83 (m, 4H, Ar)
4b	8-Cl	Н	56	47–48	_b	-	$C_{12}H_{11}CIN_2S$	2.19 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 2.58, 3.27, 3.54 (AXY system, 4H, CH <sub>2</sub> CH <sub>2</sub> ), 6.82 (d, $J = 2$ Hz, 1H, H-2), 7.17–7.79 (m, 3H, Ar)
4c	Н	CH <sub>3</sub>	42	oil	_	-	$C_{13}H_{14}N_2S$	1.41 (d, $J = 6$ Hz, 3H, SCHC $H_3$ ), 2.20 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 2.70–4.13 (m, 3H, CHCH <sub>2</sub> ), 6.80 (d, $J = 2$ Hz, 1H, H-2), 7.14–7.86 (m, 4H, Ar)
4d	8-Cl	CH <sub>3</sub>	37	120–121	EtAc	white prisms	$C_{13}H_{13}CIN_2S$	1.39 (d, $J = 6$ Hz, 3H, SCHC $H_3$ ), 2.20 (d, $J = 2$ Hz, 3H, CH $_3$ ), 2.56–4.16 (m, 3H, CHCH $_2$ ), 6.77 (d, $J = 2$ Hz, 1H, H-2), 7.06–7.80 (m, 3H, Ar)
4e	Н	C <sub>6</sub> H <sub>5</sub>	71	152–153	Acetone	white prisms	$C_{18}H_{16}N_2S$	2.25 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 2.83, 3.40, 4.83 (ABX system, 3H, CHCH <sub>2</sub> ), 6.80 (d, $J = 2$ Hz, 1H, C2-H), 7.06–7.93 (m, 9H, Ar)
<b>4f</b>	8-Cl	C <sub>6</sub> H <sub>5</sub>	47	166–167	EtAc	white prisms	C <sub>18</sub> H <sub>15</sub> ClN <sub>2</sub> S	2.23 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 2.86, 3.46, 4.86 (ABX system, 3H, CHCH <sub>2</sub> ), 6.85 (d, $J = 2$ Hz, 1H, H-2), 7.06–7.96 (m, 8H, Ar)
<b>4</b> g	Нα	≻ClC <sub>6</sub> H	I <sub>4</sub> 58	143–147	_b	-	C <sub>18</sub> H <sub>15</sub> CIN <sub>2</sub> S	2.26 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 2.83, 3.34, 5.43 (ABX system, 3H, CHCH <sub>2</sub> ), 6.89 (d, $J = 2$ Hz, 1H, H-2), 7.06–7.97 (m, 8H, Ar)
12	-	-	95	105-106	EtAc	white prisms	$C_{18}H_{14}N_2S$	2.36 (d, $J = 2$ Hz, 3H, CH <sub>3</sub> ), 6.96 (d, $J = 2$ Hz, 1H, H-2), 7.03–7.82 (m, 10H, 9Ar + 1H-4)

The 1H-NMR spectra of compounds 4c-g showed signals attributable to mixtures of isomers (60:40), as previously observed in 1-substituted-5-methyl- or 5-phenyl-4,5-dihydro-s-triazolo[3,4-d]-1,5-benzothiazepines [25]. Purified by flash chromatography and not crystallized.

orders of magnitude in receptor affinity as observed for compound 8a ( $K_i = 4.24 \mu M$ ) and the absence of the 2-position substituent (compound 4a) caused a loss of affinity. This is in agreement with the hypothesis that a lipophilic region of steric attraction is required at the 2-position [18].

The introduction of a phenyl ring at the 5-position in the benzothiazepine system resulted in a loss of

affinity, as shown for compounds 4e, 4f, 4g, 9e, 9f, 9g, 12, 15 and 16. This suggests that a phenyl group in this position may introduce a steric hindrance which would interfere with a hypothetical region of repulsive interaction for the benzodiazepine receptor [18].

The same paper [18] also suggests that antagonist/inverse agonist activities depend on the presence of a lipophilic group, a proton acceptor and a proton donor

Table II. Physical and chemical data of compounds 5a-g.

$$x = \sum_{N=1}^{R} \sum_{NH_2}^{R}$$

5a-

Compound	X	R	Yield (%)	Мр (°С)	Formula	¹H-NMR, δ
5a <sup>a</sup>	Н	Н	35	186	$C_9H_{10}N_2S$	2.55, 3.50 (2t, $A_2B_2$ system, 4H, $CH_2CH_2$ ), 5.02 (br s, 2H, $NH_2$ ), 6.70–7.53 (m, 4H, Ar)
5b	8-Cl	Н	32	131–135	C <sub>9</sub> H <sub>9</sub> ClN <sub>2</sub> S	2.52, 3.48 (2t, $A_2B_2$ system, 4H, $CH_2CH_2$ ), 5.20 (br s, 2H, $NH_2$ ), 6.90–7.50 (m, 3H, Ar)
5c	Н	CH <sub>3</sub>	41	134–138	$C_{10}H_{12}N_2S$	1.40 (d, $J = 7$ Hz, 3H, CH <sub>3</sub> ), 2.00-2.56 (m, 2H, CH <sub>2</sub> ), 3.70-4.16 (m, 1H, CH)
5d	8-Cl	CH <sub>3</sub>	49	134–137	$C_{10}H_{11}ClN_2S$	1.33 (d, $J = 6$ Hz, 3H, CH <sub>3</sub> ), 1.73–2.60 (m, 2H, CH <sub>2</sub> ), 3.20, 4.20 (superimposed br s and m, 3H, NH <sub>2</sub> and CH), 6.74–7.43 (m, 3H, Ar) (DMSO- $d_6$ )
5e	Н	C <sub>6</sub> H <sub>5</sub>	37	121–123	$C_{15}H_{14}N_2S$	2.58, 2.77, 4.93 (ABX system, 3H, CHCH <sub>2</sub> ), 4.93 (vbr s, 2H, NH <sub>2</sub> ), 6.93–7.59 (m, 9H, Ar)
5f	8-Cl	C <sub>6</sub> H <sub>5</sub>	27	131–133	$C_{15}H_{13}CIN_2S$	2.50–2.85 (m, 2H, CH <sub>2</sub> ), 4.56–5.23 (superimposed br s and m, 3H, NH <sub>2</sub> and CH), 6.73–7.86 (m, 8H, Ar) (DMSO- $d_6$ )
5g	На	o-ClC <sub>6</sub> l	I <sub>5</sub> 50	125–127	$C_{15}H_{13}ClN_2S$	2.76–3.08 (m, 2H, CH <sub>2</sub> ), 4.23 (vbr s, 2H, NH <sub>2</sub> ), 5.20–5.53 (m, 1H, CH), 7.00–7.60 (m, 8H, Ar) (DMSO- $d_6$ )

<sup>&</sup>lt;sup>a</sup>Known [3], mp 175-177°C, yield 77%.

group. Another recent article [19] reported the synthesis of a new benzodiazepine ligand which had no proton donor group. Both this benzodiazepine ligand and 9a have a lipophilic group (phenyl) and a proton acceptor (nitrogen at 3-position) in the five-membered ring and are devoid of proton donor group.

These remarks suggest that the presence of a proton donor group is not an essential requirement for the antagonist/inverse agonist activities. The GABA-ratio value of 9a and its structural requirement analogy with other compounds [18, 19] suggest that it could act as an antagonist or an inverse agonist. This compound will be submitted to further investigations to define its pharmacological profile.

#### **Experimental protocols**

#### Chemistry

Melting points were taken on a Kofler hot-stage apparatus and are uncorrected.  $^1H\text{-}NMR$  spectra were recorded in  $CDCl_3$  solution, unless otherwise indicated, using a Varian Em-390 (90 MHz) spectrometer. The chemical shift values are reported in  $\delta$  (ppm) relative to tetramethylsilane as an internal standard.

Mass spectra were recorded on a Varian MAT 311A spectrometer. Elemental analyses were performed for C, H and N on a Carlo Erba Elemental Analyzer model 1106 and results were within  $\pm 0.4\%$  of the theoretical values. The purity of the compounds was checked by TLC (pre-coated silica-gel plates, Merck Kieselgel 60  $F_{254}$ ). Flash chromatographies were performed on columns packed with Merck silica gel, 230–400 mesh.

General procedure for 1-methylimidazo[2,1-d][1,5]benzothiazepines 4a-g and 12

A mixture of 2a-d or 3 (1 mmol), propargylamine hydrochloride (1.1 mmol) and sodium acetate (1.1 mmol) in *n*-butanol (10 ml) for 2a-d or anhydrous ethanol (10 ml) for 3 was refluxed under a nitrogen current for 15 h for 4a, 4c and 4d, 5 h for 4b and 4e, and 10 h for 4f, 4g and 12. The reaction mixture was then evaporated in vacuo and the residue was dissolved in chloroform and extracted with diluted hydrochloric acid solution. The acidic layer was neutralized with 30% ammonium hydroxide solution and extracted with chloroform. Finally, the chloroform phase was washed with water, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and evaporated in vacuo. The residue was either induced to crystallize by adding small amounts of hexane/ethyl acetate 1:1 for 4a, ethyl acetate for 4d and acetone for 4e, or purified by flash chromatography (chloroform) (table 1).

Table III. Physical and chemical data of compounds 6a-g, 7a-g, 13 and 14.

6a.o. 7a.o

13,14

Compound	X	R	$R^{1}$	Yield (%)	Mp (°C)	Crystallization solvent	n Formula	¹H-NMR, δ
ба	Н	Н	CH <sub>3</sub>	70	103–105 dec	EtOH	C <sub>12</sub> H <sub>13</sub> NO <sub>2</sub> S	2.29 (s, 3H, CH <sub>3</sub> ), 2.63 (t, 2H, CH <sub>2</sub> ) 3.38 (br t, 2H, CH <sub>2</sub> ), 4.06, 4.95 (2br d 2H, NCH <sub>2</sub> ), 7.13–7.68 (m, 4H, Ar)
6b	8-Cl	Н	CH <sub>3</sub>	26	94–96	EtAc	C <sub>12</sub> H <sub>12</sub> CINO <sub>2</sub> S	2.28 (s, 3H, COCH <sub>3</sub> ), 2.65 (t, 2H, CH <sub>2</sub> ) 3.40 (br t, 2H, CH <sub>2</sub> ), 3.97, 4.96 (2 br d 2H, NCH <sub>2</sub> ), 7.10–7.64 (m, 3H, Ar)
6c	Н	CH <sub>3</sub>	CH <sub>3</sub>	81	50–55	a	C <sub>13</sub> H <sub>15</sub> NO <sub>2</sub> S	1.43 (br d, 3H, CHCH <sub>3</sub> ), 2.32 (s, 3H, COCH <sub>3</sub> ), 2.40–2.95 (m, 2H, CHCH <sub>2</sub> ), 3.65–4.05 (m, 1H, CHCH <sub>2</sub> ), 4.25, 4.85 (2 br d, 2H, NCH <sub>2</sub> ), 7.10–7.73 (m, 4H, Ar)
6d	8-CI	СН3	CH <sub>3</sub>	100	129–132	EtAc	C <sub>13</sub> H <sub>14</sub> CINO <sub>2</sub> S	1.40 (br d, 3H, CHCH <sub>3</sub> ), 2.29 (s, 3H, COCH <sub>3</sub> ), 2.38–2.90 (m, 2H, CHCH <sub>2</sub> ), 3.53–4.10 (m, 1H, CHCH <sub>2</sub> ), 4.06, 4.90 (2 br d, 2H, NCH <sub>2</sub> ), 7.03–7.63 (m, 3H, Ar)
6e	Н	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	75	89–90	EtAc	C <sub>18</sub> H <sub>17</sub> NO <sub>2</sub> S	2.32 (s, 3H, COCH <sub>3</sub> ), 2.70–3.05 (m, 2H, CHC $H_2$ ), 4.06, 5.00 (AX system, $J = 18$ Hz, 2H, NCH <sub>2</sub> ), 4.56–5.06 (m, 1H, C $H$ CH <sub>2</sub> ), 7.03–7.70 (m, 9H, Ar)
6f	8-Cl	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	51	152–154	EtAc	C <sub>18</sub> H <sub>16</sub> ClNO <sub>2</sub> S	2.30 (s, 3H, COCH <sub>3</sub> ), 2.74–3.04 (m, 2H, CHC $H_2$ ), 4.03, 5.02 (AX system, $J = 17.6$ Hz, 2H, NCH <sub>2</sub> ), 4.74–4.90 (m, 1H, CHCH <sub>2</sub> ), 7.06–7.78 (m, 8H, Ar)
6g	На	o-ClC <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	90	103–105	EtAc	C <sub>18</sub> H <sub>16</sub> ClNO <sub>2</sub> S	2.55 (s, 3H, COCH <sub>3</sub> ), 2.96-3.25 (m, 2H, CHCH <sub>2</sub> ), 4.36, 5.24 (AX system, $J = 17.6$ Hz, NCH <sub>2</sub> ), 5.53-5.68 (m, 1H, CHCH <sub>2</sub> ), 7.25-7.82 (m, 8H, Ar)
7a	Н	Н	C <sub>6</sub> H <sub>5</sub>	57	85–87	EtAc	$C_{17}H_{15}NO_2S$	2.66, 3.36 (2t, $A_2B_2$ system, $J=6$ Hz, 4H, $CH_2CH_2$ ), 4.60, 5.63 (2br d, 2H, $CH_2$ ), 6.90–8.00 (m, 9H, Ar)
7b	8-Cl	Н	C <sub>6</sub> H <sub>5</sub>	80	134–135	EtAc	C <sub>17</sub> H <sub>14</sub> CINO <sub>2</sub> S	2.70 (br t, 2H, CH <sub>2</sub> ), 3.42 (br t, 2H, CH <sub>2</sub> ), 4.48, 5.76 (2 vbr d, 2H, NCH <sub>2</sub> ), 7.13–8.10 (m, 8H, Ar)
7c	Н	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	70	119–121	EtOH	C <sub>18</sub> H <sub>17</sub> NO <sub>2</sub> S	1.36 (br d, 3H, CH <sub>3</sub> ), 1.93–2.86 (m, 2H, CHCH <sub>2</sub> ), 3.60–4.06 (m, 1H, CH-CH <sub>2</sub> ), 4.57, 5.76 (AX system, <i>J</i> = 17 Hz, 2H, NCH <sub>2</sub> ), 6.60–8.50 (m, 9H, Ar)
7 <b>d</b>	8-Cl	CH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub>	75	109–110	EtOH	C <sub>18</sub> H <sub>16</sub> CINO <sub>2</sub> S	1.35 (d, $J$ = 6 Hz, 3H, CH <sub>3</sub> ), 2.30–2.75 (m, 2H, CHC $H_2$ ), 3.67–4.04 (m, 1H, CHCH <sub>2</sub> ), 4.50, 5.28 (AX system, $J$ = 17.5 Hz, 2H, NCH <sub>2</sub> ), 7.12–8.10 (m, 8H, Ar)

Table III. (Continued.)

Compound	X	R	$R^{I}$	Yield (%)	Mp (°C)	Crystallization solvent	Formula	¹H-NMR, δ
7e	Н	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	43	138–139	EtOH	C <sub>23</sub> H <sub>19</sub> NO <sub>2</sub> S	2.80-3.15 (m, 2H, CHCH <sub>2</sub> ), 4.16, 5.87 (AX system, J = 17 Hz, 2H, NCH <sub>2</sub> ), 4.78-4.95 (m, 1H, CHCH <sub>2</sub> ), 7.13-8.12 (m, 14H, Ar)
7f	8-Cl	C <sub>6</sub> H <sub>5</sub>	C <sub>6</sub> H <sub>5</sub>	28	128–130	EtAc/ cyclohexane 1:1	C <sub>23</sub> H <sub>18</sub> CINO <sub>2</sub> S	2.80–3.15 (m, 2H, CHC $H_2$ ), 4.60, 5.80 (AX system, $J = 17.5$ Hz, 2H, NCH <sub>2</sub> ), 4.75–4.95 (m, 1H, CHCH <sub>2</sub> ), 7.12–8.15 (m, 13H, Ar)
7g	Нα	o-ClC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	64	128–129	EtOH	C <sub>23</sub> H <sub>18</sub> CINO <sub>2</sub> S	2.77-3.14 (m, 2H, CHC $H_2$ ), 4.65, 5.88 (AX system, $J = 17.6$ Hz, 2H, NCH <sub>2</sub> ), 5.34-5.50 (m, 1H, C $H$ CH <sub>2</sub> ), 7.05-8.14 (m, 13H, Ar)
13	-	_	CH <sub>3</sub>	90	119–120	EtOH	$C_{18}H_{15}NO_2S$	2.33 (s, 3H, COCH <sub>3</sub> ), 4.66 (1br d, 2H, NCH <sub>2</sub> ), 6.45 (s, 1H, H-3), 7.03–7.86 (m, 9H, Ar)
14		-	C <sub>6</sub> H <sub>5</sub>	90	138-140	EtAc	$C_{23}H_{17}NO_2S$	4.94, 5.73 (AX system, $J = 17.2$ Hz, 2H, NCH <sub>2</sub> ), 6.53 (s, 1H, H-3), 7.10–8.20 (m, 14H, Ar)

<sup>&</sup>lt;sup>a</sup>Purified by flash chromatography and not crystallized.

General procedure for 2-methyl- and 2-phenylimidazo[2,1-d]-[1,5]benzothiazepines 8 and 9

Compounds 8 and 9 were synthesized according to two methods. *Method A* starting from 2,3-dihydrobenzothiazepin-3(5H)-thiones 2 and *Method B* starting from 2,3-dihydro-1,5-benzothiazepin-3(5H)-ones 1 and from 2-phenyl-1,5-benzothiazepin-3(5H)-one 10.

General procedure for 4-amino-2,3-dihydro-1,5-benzothiaze-pines 5a-g. Method A. A mixture of 2,3-dihydrobenzothiaze-pin-3(5H)-thione 2a-g (10 mmol) in methanol saturated with ammonia (50 ml) was stirred at room temperature for 20 h. The mixture was then filtered and the filtrate was concentrated to half its volume, cooled in ice and the resulting precipitate filtered to give 5a and 5e-g. In the other cases, the filtrate was brought to dryness in vacuo and induced to crystallize by adding small amounts of ethyl acetate or methanol (5c, d) or was purified by flash chromatography (chloroform/methanol 95:5) (5b). These compounds were used without further purification (table II).

2-Methyl-imidazo[2,1-d][1,5]henzothiazepines 8c and 8d. Method A. A mixture of 4-aminobenzothiazepine 5c and 5d (1 mmol), chloroacetone (1 mmol) and sodium acetate (1 mmol) in anhydrous ethanol (10 ml) was refluxed for 16 h (8c) or 20 h (8d). After cooling the mixture was filtered and the filtrate evaporated in vacuo. The residue was dissolved with chloroform, washed with water, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, filtered and dried. The residue was purified by flash chromatography (chloroform) (table IV).

General procedure for 2-phenyl-imidazo[2,1-d][1,5]benzothiazepines 9c-e and 9g. Method A. A mixture of 4-aminobenzothiazepines 5c-e and 5g (1 mmol), phenacyl bromide (1 mmol) and sodium bicarbonate (1 mmol) in anhydrous ethanol (10 ml) was stirred for 45 min at room temperature and was then refluxed for 1.5 h. The reaction mixture was evaporated in vacuo and the residue was dissolved with actione and filtered. The filtrate was brought to dryness and the residue was either crystallized by adding small amounts of hexane (9c), or ethyl acetate (9d and 9g) or purified by flash chromatography (methylene chloride) (9e) (table V).

General procedure for N-(propan-2'-one)-1,5-benzothiazepin-2(5H)-ones 6a-g and 13 and N-(phenacyl)-1,5-benzothiazepin-2(5H)-ones 7a-g and 14. Method B. To a stirred solution of the benzothiazepine derivatives 1 or 10 [1] (1 mmol), tetrabutylammonium bromide (0.1 mmol) and chloroacetone (1.5 mmol) or phenacylbromide (1.5 mmol) in tetrahydrofuran (ca 10 ml), finely powdered potassium hydroxide (1.5 mmol) was added. The reaction mixture was kept at room temperature under stirring for 24-72 h for compounds 6a-g and 13, 6 h for 5b and 16-40 h for compounds 5a, 5c-g and 14 and then filtered. The filtrate was evaporated in vacuo and the residue taken up with chloroform, the chloroform extract washed with water, dried over Na<sub>2</sub>SO<sub>4</sub> and brought to dryness in vacuo. The crude product was then either crystallized by adding small amounts of ethanol or ethyl acetate (6d, 6f, 6g, 7c-e, 7g and 13), or purified by flash chromatography (chloroform) (6a-c, 6e, 7a, 7b, 7f and 14) (table III).

General procedure for 2-methyl- and 2-phenylimidazol[2,1-d]-[1,5]benzothiazepines 8a-g, 9a-c, 9f, 15 and 16. Method B. A mixture of N-alkyl derivative 6a-g, 7a-c, 7f, 13 and 14 (1 mmol) and ammonium acetate (10 mmol) in glacial acetic acid (10 ml), was refluxed for 3 h for 8a, 8b and 16, for 6 h for 8c, 8e, 9b, 9c, 9f and 15 and 10 h for 8d, 8f, 8g and 9a. After cooling the solution was poured into ice-water and neutralized

Table IV. Physical and chemical data of compounds 8a-g and 15.

Compound X R Yield (%)  $^{1}H$ -NMR,  $\delta$ Crystal-Formula Mр  $(^{\circ}\bar{C})$ lization Method A Method B solvent 8a Н Η 37 119-121 **EtOH** 2.26 (s, 3H, CH<sub>3</sub>), 2.93, 3.42 (A<sub>2</sub>B<sub>2</sub>  $C_{12}H_{12}N_2S$ system, 4H, CH<sub>2</sub>CH<sub>2</sub>), 6.76 (s, 1H, H-1), 7.03–7.73 (m, 4H, Ar) 8ь 8-Cl H 62 145-147 EtAc C<sub>12</sub>H<sub>11</sub>ClN<sub>2</sub>S 2.30 (s, 3H,  $CH_3$ ), 3.00, 3.50 ( $A_2B_2$ system, 4H, CH<sub>2</sub>CH<sub>2</sub>), 6.80 (s, 1H, H-1), 7.42–7.73 (m, 4H, Ar) Н 1.43 (d, J = 5 Hz, 3H, SCHC $H_3$ ), 8c  $CH_3$ 46 65 47-49 EtAc  $C_{13}H_{14}N_2S$ 2.29 (s, 3H, CH<sub>3</sub>), 2.56, 3.04, 3.85 (ABC system, 3H, CHCH<sub>2</sub>), 6.76 (s, 1H, H-1), 6.84-7.73 (m, 4H, 1.40 (d, J = 5 Hz, 3H, SCHC $H_3$ ), 2.30 (s, 3H, CH<sub>3</sub>), 2.60, 3.03, 3.91 (ABC system, 3H, CHCH<sub>2</sub>), 6.80 (s, 1H, H-1), 7.10–7.75 (m, 3H, Ar) 8d 8-C1 CH<sub>3</sub> 57 60 50-52  $C_{13}H_{13}ClN_2S$ 8e Η C<sub>6</sub>H<sub>5</sub> 30 115-116 2.32 (s, 3H, CH<sub>3</sub>), 3.00, 3.40, 4.93 **EtAc**  $C_{18}H_{16}N_2S$ (ABX system, 3H, CHCH<sub>2</sub>), 6.90 (s, 1H,  $C_1$ -H), 7.00–7.86 (m, 9H, Ar) 8-Cl C<sub>6</sub>H<sub>5</sub> 8f 46 168 Acetone  $C_{18}H_{15}ClN_2S$ 2.36 (s, 3H, CH<sub>3</sub>), 3.03, 3.36, 4.91 (ABX system, 3H, CHCH<sub>2</sub>), 6.89 (s, 1H, H-1), 7.20–7.87 (m, 8H, Ar) 2.31 (s, 3H, CH<sub>3</sub>), 2.89, 3.37, 5.41 (ABX system, 3H, CHCH<sub>2</sub>), 6.84 8g H o-ClC<sub>6</sub>H<sub>4</sub> 31 102-104 EtAc C<sub>18</sub>H<sub>15</sub>ClN<sub>2</sub>S (s, 1H, H-1), 6.95-7.75 (m, 8H, Ar) 15 2.38 (s, 3H, CH<sub>3</sub>), 7.11 (s, 1H, H-1), 37 108-110 EtAc/  $C_{18}H_{14}N_2S$ cyclo-7.22-7.88 (m, 10 H, Ar) hexane 1:1

with 30% ammonium hydroxide solution. In the case of 8f, 9a, 9c and 9f, the resulting precipitate was filtered and recrystallized from a suitable solvent. In the other cases, the mixture was extracted with chloroform. The chloroform solution was washed with water, dried over Na<sub>2</sub>SO<sub>4</sub>, filtered and brought to dryness in vacuo. The oily residue was either induced to crystallize by adding small amounts of ethyl acetate or ethanol, or purified by flash chromatography (chloroform) (tables IV and V).

#### Receptor-binding assay

Tritiated flunitrazepam was obtained from Du Pont de Nemours New England Nuclear Division (Dreieichenhaim, Germany) and had a specific activity of 83.5 Ci/mmol and a radiochemical purity > 99%. All other chemicals were reagent grade and obtained from commercial suppliers.

The [ $^{3}$ H]Flunitrazepam binding assay to bovine cerebral cortex membrane was carried out essentially as described previously [20]. Cortices were rapidly isolated and homogenized in 10 volumes of ice-cold 0.32 M sucrose containing protease inhibitor [21]. The homogenate was centrifuged at 1000 g for 15 min at 4°C. The pellet was then osmotically shocked by suspension in 10 vol of 50 mM Tris-HCl buffer at pH 7.4 containing protease inhibitors and recentrifuged at 5000 g for 15 min at 4°C. The resulting membranes

<sup>\*</sup>Purified by flash chromatography and not crystallized.

Table V. Physical and chemica data of compounds 9a-g and 16.

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1-2			

Compound	X	R	Yield	i (%)	Mp	Crystal- Formula	¹H-NMR, δ
			Method A	Method B	- (°C)	lization solvent	
9a <sup>a</sup>	Н	Н		67	118-119	Acetone C <sub>17</sub> H <sub>14</sub> N <sub>2</sub> S	3.08, 3.53 (A <sub>2</sub> B <sub>2</sub> system, 4H, CH <sub>2</sub> · CH <sub>2</sub> ), 6.76–8.00 (m, 10H, Ar + H-1)
9b	8-Cl	Н		46	88–90	EtAc C <sub>17</sub> H <sub>13</sub> ClN <sub>2</sub> S	3.09, 3.53 (A <sub>2</sub> B <sub>2</sub> system, 4H, CH <sub>2</sub> -CH <sub>2</sub> ), 7.20–7.88 (m, 9H, Ar + H-1)
9c	H	CH <sub>3</sub>	61	90	170–171	Acetone C <sub>18</sub> H <sub>16</sub> N <sub>2</sub> S	1.46 (d, $J = 6$ Hz, 3H, CH <sub>3</sub> ), 2.63, 3.20, 3.95 (ABC system, 3H, CH-CH <sub>2</sub> ), 7.06–7.92 (m, 9H, Ar + H-1)
9d	8-Cl	CH <sub>3</sub>	70		151–152	EtAc C <sub>18</sub> H <sub>15</sub> ClN <sub>2</sub> S	1.46 (d, $J = 6$ Hz, 3H, CH <sub>3</sub> ), 2.63, 3.16, 3.96 (ABC system, 3H, CH-CH <sub>2</sub> ), 7.10–7.93 (m, 9H, Ar + H-1)
9e	Н	C <sub>6</sub> H <sub>5</sub>	47		117–120	$-b$ $C_{23}H_{18}N_2S$	3.10, 3.50, 4.93 (ABX system, 3H, CHCH <sub>2</sub> ), 6.80-7.96 (m, 15H, Ar + H-1)
9f	8-Cl	C <sub>6</sub> H <sub>5</sub>		80	164166	Acetone C <sub>23</sub> H <sub>17</sub> ClN <sub>2</sub> S	2.97, 3.48, 4.93 (ABX system, 3H, CHCH <sub>2</sub> ), 7.13-7.90 (m, 14H, Ar + H-1)
9g	Но	-ClC <sub>6</sub> H <sub>4</sub>	23		176–185	EtAc C <sub>23</sub> H <sub>17</sub> ClN <sub>2</sub> S	2.80, 3.23, 5.26 (ABX system, 3H, CHCH <sub>2</sub> ), 6.90-7.64 (m, 14H, Ar)
16				51	85–86	EtAc C <sub>23</sub> H <sub>16</sub> N <sub>2</sub> S	7.20–8.10 (m, 16H, 15Ar + H-1)

<sup>&</sup>lt;sup>a</sup>Known [3], mp 119-120°C, yield 57%. <sup>b</sup>Purified by flash chromatography and not crystallized.

were frozen and washed using a procedure previously described for removing endogenous GABA from rat cerebral cortex [22]. Finally, the pellet was suspended in 10 volumes of 50 mM Tris-HCl buffer at pH 7.4 and used in the binding assay.

Protein concentration was assayed by the method of Lowry et al [23].

Routine [3H]Flunitrazepam binding assays were run by incubating 0.4 mg protein of crude bovine brain membrane suspension at 0°C for 90 min with [3H]Flunitrazepam (0.2 nM) in a total volume of 0.5 ml of Tris-HCl buffer. After incubation, samples were diluted at 0°C with 5 ml of the assay buffer and immediately filtered under reduced pressure through glass-fiber filter disks (Whatman GF/B). The filters were washed with 5 ml of the buffer, dried and added to 8 ml of Ready Protein

Beckman scintillation counter and radioactivity was counted on a LS 1800 scintillation counter. Non-specific binding was determined by the radioactivity bound in the presence of  $10~\mu M$  non-radioactive diazepam in parallel assays. Water-insoluble derivatives were dissolved in EtOH and added to the assay mixture. Blank experiments were carried out to determine the effect of EtOH (2%) or DMSO (1%) on binding. Values of inhibition of specific binding by 50% (IC<sub>50</sub>) were calculated from the displacement curves by log-probit analysis with four to six concentrations of the displacers, each performed in triplicate.

Dissociation constants  $K_i$  were derived according to the equation of Chen and Prusoff [24]. The ligand affinity  $K_d$  of [3H]Flunitrazepam was 1.8 nM.

Table VI. Inhibition of [3H]Flunitrazepam binding by compounds 4a-g, 8a-g, 12, 15 and 16.

Compound	X	R	$R_{I}$	Inhibition %² (10 μM)		Compound	X	R	$R_1$	Inhibition %a (10 μΜ		GABA- ratio
4a	Н	Н	_	34.4	_	8e	Н	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	16.9	_	_
4b	8-Cl	H	_	16.4	_	8f	8-Cl	$C_6H_5$	CH <sub>3</sub>	20.97	-	_
4c	H	CH <sub>3</sub>	_	33.2	_	8g	Н	o-ClC <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	30.65		_
4d	8-Cl	CH <sub>3</sub>	_	33.2	_	15	_	-	$CH_3$	16.8	-	_
4e	Н	$C_6H_5$	-	45.34	_	9a	Н	Н	$C_6H_5$	98.9	0.043 ±0.00	211.09
4f	8-Cl	$C_6H_5$	_	45.34	-	9b	8-Cl	H	$C_6H_5$	52	_	_
4g	Н	o-ClC <sub>6</sub> H <sub>4</sub>	-	42.9	-	9c	Н	$CH_3$	$C_6H_5$	94.79	$0.097 \pm 0.00$	730.8
12	_		_	2.2	_	9 <b>d</b>	8-C1	$CH_3$	$C_6H_5$	81.3	$1.97 \pm 0.15$	5 –
8a	Н	Н	$CH_3$	63.1 4	1.24 ± 0.3	2 <b>9e</b>	Н	$C_6H_5$	$C_6H_5$	0	-	<del>-</del>
8b	8-Cl	Н	$CH_3$	83.47 1	$.33 \pm 0.0$	98 <b>9f</b>	8-Cl	$C_6H_5$	$C_6H_5$	0	-	-
8c	Н	CH <sub>3</sub>	$CH_3$	59.04 5	5.34 ± 0.4	3 <b>9g</b>	Н	o-ClC <sub>6</sub> H <sub>4</sub>	$C_6H_5$	0	-	-
8d	8-Cl	CH <sub>3</sub>	$CH_3$	32.23	_	16	_	_	$C_6H_5$	0	-	-
Chlordiazep	ooxide			0	0.72 ± 0.0	7						1.73

<sup>a</sup>Percent of inhibition of specific [<sup>3</sup>H]Flunitrazepam binding at 10 μM compound concentration; results are the means of three separate experiments with SEM less than 10%. <sup>b</sup>Means ± SEM of three determinations.

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